



| 理論化学、計算化学とは?                                                                     |                                                                                                                 |                                                                                                |  |  |  |  |
|----------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------|--|--|--|--|
| 理論化学、計算化学:                                                                       |                                                                                                                 |                                                                                                |  |  |  |  |
| 理論(モデルや概念)の<br>ご 既知の実験事実を                                                        | D構築や<br>説明、未知                                                                                                   | その適用(計算手法の開発、数値計算、紙と鉛筆)<br>Elの物質の性質などを予言                                                       |  |  |  |  |
| 実験化学:                                                                            |                                                                                                                 |                                                                                                |  |  |  |  |
| <b>多数の実験</b>                                                                     | 皆後にある                                                                                                           | る普遍的な理論を導く                                                                                     |  |  |  |  |
| 対象:物質の構造                                                                         | 、物質の                                                                                                            | の性質、物質の反応                                                                                      |  |  |  |  |
| 小サイズの単分子<br>弱電子相関系<br>基底状態<br>時間無依存<br>量子系、古典系<br>定常、平衡状態<br>均一系<br>温度無依存<br>古典場 | $\rightarrow$ $\rightarrow$ $\rightarrow$ $\rightarrow$ $\rightarrow$ $\rightarrow$ $\rightarrow$ $\rightarrow$ | 巨大分子、分子集合系<br>電子相関系<br>励起状態<br>時間発展(ダイナミクス)<br>量子-古典ハイブッリド<br>非定常、非平衡状態<br>不均一系<br>温度効果<br>量子場 |  |  |  |  |
|                                                                                  |                                                                                                                 |                                                                                                |  |  |  |  |









| Unique Pro                                                                                                                      | operties of Open-Shell Singlet Molecular Systems                                                                                                                                                                                                                                                                |  |  |  |
|---------------------------------------------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--|--|--|
| • Longer acenes possess a nonzero bandgap with a singlet open-<br>shell ground state followed by a triplet state a few kcal/mol |                                                                                                                                                                                                                                                                                                                 |  |  |  |
| nigner in                                                                                                                       | M. Bendikov et al. J. Am. Chem. Soc. <b>126</b> , 7416 (2004).                                                                                                                                                                                                                                                  |  |  |  |
| • High rea                                                                                                                      | • High reactivity of the zigzag edges of the longer oligoacenes.                                                                                                                                                                                                                                                |  |  |  |
| High cha     Electron                                                                                                           | arge carrier mobilities in the solid state.<br>ic states are characterized by "diradical character".                                                                                                                                                                                                            |  |  |  |
| K. Yamagua<br>Theory and                                                                                                        | chi, T. Fueno, H. Fukutome, <i>Chem. Phys. Lett.</i> 22, 461 (1973); in, <i>Self-Consistent Field:</i><br><i>Applications</i> , R. Carbo et al. Eds.; Elsevier: Amsterdam, pp. 727 (1990).                                                                                                                      |  |  |  |
| • Excitation<br>diradcial<br>M. Nakano                                                                                          | on energies and properties are strongly correlated to the<br>l character in the ground state.<br>et al. <i>Phys. Rev. Lett.</i> <b>99</b> , 033001 (2007)                                                                                                                                                       |  |  |  |
| • Diradcia<br>optical r<br>M. Nakano<br>T. Minami,                                                                              | <b>I character control of optical properties, e.g., nonlinear</b><br><b>responses, optical absorption, singlet fission, etc.</b><br>et al., <i>J. Phys. Chem. A</i> <b>109</b> , 885 (2005); <i>J. Chem. Phys.</i> <b>133</b> , 154302 (2010);<br>M. Nakano, <i>J. Phys. Chem. Lett.</i> <b>3</b> , 145 (2012). |  |  |  |







































| Two-site model A <sup>-</sup> –B <sup>·</sup> with two electrons in two orbitals:<br>Valence configuration interaction (VCI) approach                                                                                                                                                       |  |  |  |  |  |  |
|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--|--|--|--|--|--|
| Symmetry adapted MOs $\dots g(x), u(x)$                                                                                                                                                                                                                                                     |  |  |  |  |  |  |
| <b>Localized natural orbitals (LNOs)</b> $a(x), b(x)  \langle a b \rangle = 0$                                                                                                                                                                                                              |  |  |  |  |  |  |
| $a(x) = \frac{1}{\sqrt{2}} [g(x) + u(x)] \qquad b(x) = \frac{1}{\sqrt{2}} [g(x) - u(x)]$                                                                                                                                                                                                    |  |  |  |  |  |  |
| Basis sets for $M_s = 0$ (singlet and triplet) states<br>$\{ a\bar{b}\rangle,  b\bar{a}\rangle,  a\bar{a}\rangle,  b\bar{b}\rangle\}$ , where $ a\bar{b}\rangle =  core  a\bar{b}\rangle$ ,                                                                                                 |  |  |  |  |  |  |
| CI matrix using LNOs                                                                                                                                                                                                                                                                        |  |  |  |  |  |  |
| $ \begin{pmatrix} 0 & K_{ab} & t_{ab} & t_{ab} \\ K_{ab} & 0 & t_{ab} & t_{ab} \\ t_{ab} & t_{ab} & U & K_{ab} \\ t_{ab} & t_{ab} & K_{ab} & U \end{pmatrix} \qquad \qquad$ |  |  |  |  |  |  |





















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| Poly | cyclic di         | phenale       | nyl radio                        | cals: con                   | trol of di | radical c           | haracter |
|------|-------------------|---------------|----------------------------------|-----------------------------|------------|---------------------|----------|
| PERO | y = 0.2620        | α = 605 a.u.  | $\gamma = 51 \times 10^3 a.u.$   | Aromaticity of middle rings |            | Diradical character |          |
| PDPL | <i>y</i> = 0.5833 | α = 744 a.u.  | $\gamma = 1255 \times 10^3 a.u.$ |                             | Small      | •                   | Small    |
|      | Ć                 |               |                                  |                             |            |                     |          |
| IDPL | <i>y</i> = 0.7461 | α = 896 a.u.  | $\gamma = 2383 \times 10^3 a.u.$ |                             |            |                     |          |
|      | Ś                 |               | >                                |                             |            |                     |          |
| NDPL | <i>y</i> = 0.8317 | α = 1115 a.u. | $\gamma = 3803 \times 10^3 a.u.$ |                             |            |                     |          |
|      |                   |               |                                  |                             |            |                     |          |
| ADPL | y = 0.8821        | α = 1380 a.u. | $\gamma = 5935 \times 10^3 a.u.$ |                             |            |                     |          |
|      |                   | $\infty$      |                                  | LYP/6-31G*                  | Large      | •                   | Large    |































































| Summary                                                                                                                                                                                                                                                                                                                                           |
|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| • <b>Diradical character</b> <i>y</i> (= <b>Bond weakness</b> = <b>Electron correlation</b> ) is correlated with the excitation energies and properties.                                                                                                                                                                                          |
| • Open-shell singlet systems with intermediate diradical charcater<br>exhibit enhanced second hyperpolarizaibilities (γ) as compared to<br>conventional closed-shell and pure diradical systems.                                                                                                                                                  |
| <ul> <li>Control of diradical character (in the ground state):<br/>Relation to conventional chemical concepts and indices</li> <li>Edge shape, size, and architecture of graphene nanoflakes</li> <li>Quinoid (closed-shell) and benzenoid (open-shell) resonance structures</li> <li>Aromaticity, Multiple bond, Main group elements,</li> </ul> |
| • Examples of diradical/multiradical<br>Diphenalenyl compounds, Geaphene nanoflakes,<br>Open-shell aggregates, etc.                                                                                                                                                                                                                               |
| These results demonstrate the high potential of open-shell singlet molecular systems for third-order NLO applications.                                                                                                                                                                                                                            |







